

Synthesis of Some Novel Chromenopyrimidine Derivatives and Evaluation of Their Biological Activities

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Abstract

Pyrimidine nucleosides are constituents of fundamental structure of the cells. There has been considerable attentions in the chemistry of pyrimidine derivatives due to having a wide range of biological activities such as antiviral, anti-malarial agents, cytostatic, antithelemintic, antibacterial, adenosine receptor ligands, anti-cancer agents, compounds targeting delayed-type hypersensitivity and anti-convulsant agents. As a part of our research work in the synthesis of pyrimidines containing biological activities, a series of chromenopyrimidine derivatives were synthesized by reaction of an intermediate imine and ammonia derivatives in good to high yields. All synthesized compounds were characterized using IR and NMR (¹H and ¹³C) spectroscopy and elemental analysis data. The antibacterial activity of these compounds was investigated against *Staphylococcus aureus* (RTCC, 1885), and *Escherichia Coli* (ATCC, 35922).

Keywords: Imine; Chromenopyrimidine; Pyranopyrimidine; Antibacterial; Multicomponent.

Introduction

Pyrimidine is a basic nucleus in DNA and RNA and plays an essential role in chemistry and biological systems (1). Pyrimidine derivatives have received considerable attentions due to their diverse range of therapeutic and pharmacological properties as antiviral (2), cytostatic (3-5), immunomodulating and antibacterial (6-9). Fused heterocyclic pyrimidines have also shown a wide range of biological activities, such as antitumor, antiviral, antimicrobial, antibacterial and anti-inflammatory (10-17). Study of various substituted pyrimidine derivatives indicated a good correlation between compound structures and antibacterial activity (18-24). Several methods have been reported for the synthesis

of simple pyrimidine derivatives (1, 25-28). However, because of the incessant interest in this field, new efficient synthesis of some fused pyrimidine derivatives is still an important objective for synthetic organic chemists in order to find compounds with different biological activities.

In view of these reports and also due to continuation of our interests on synthesis of pyrimidines (29-32), we wish to report synthesis of some fused pyrimidine derivatives.

Experimental

Materials

All reagents and solvents used are commercially available. Reactions were monitored by thin layer chromatography (TLC) using silica gel F₂₅₄ aluminum sheets (Merck). Melting points were measured on an

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Electrothermal apparatus. Infra-red spectra were recorded (KBr discs) with a Galaxy Series FT-IR 5000 spectrometer. The ^1H NMR and ^{13}C NMR spectra were recorded with a Bruker Avance 300 MHz spectrometer with $\text{DMSO-}d_6$ and CDCl_3 as the solvent and tetramethylsilane as an internal standard. Microanalyses were performed by the Elemental Analyzer (Elemental, Vario EL III) at the Arak University. The microbial strains are identified strains and were obtained from the Pasteur Institute of Iran. The bacterial strains studied are *Staphylococcus aureus* (RTCC, 1885), and *Escherichia Coli* (ATCC, 35922).

General procedure for the synthesis of compounds 6-8

In a typical experimental procedure (34-36), benzaldehyde 1 (1 mmol), malononitrile 2 (1 mmol), dimedone 3, resorcinol 4 or 2-naphthol 5 (1 mmol) were mixed in solvent and triethylamine (2-3 drops) as a catalyst was added. The reaction mixture was refluxed for 2-4 h. After the completion of the reaction, it was filtered and recrystallized from ethanol to afford the pure product 6-8.

2-Amino-7,7-dimethyl-5-oxo-4-phenyl-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (6): Yield: 85% m.p 227-229 °C. IR (KBr): 3395, 3326, 2199, 1682 cm^{-1} . Elemental analysis. Found, %: C 73.52; H 6.28; N 9.36, $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}_2$, Calculated, %: C 73.45; H 6.16; N 9.52.

2-Amino-7-hydroxy-4-phenyl-4H-chromene-3-carbonitrile (7): Yield: 78%, m.p 234-235 °C. IR (KBr): 3499, 3427, 3331, 2193 cm^{-1} . Elemental analysis. Found, %: C 72.48; H 4.61; N 10.7. $\text{C}_{16}\text{H}_{12}\text{N}_2\text{O}_2$, Calculated, %: C 72.72; H 4.58; N 10.60.

3-Amino-1-phenyl-1H-benzo[f]chromene-2-carbonitrile (8): Yield: 80%, m.p 279-280 °C. IR (KBr): 3435, 3338, 2183 cm^{-1} . Elemental analysis. Found, %: C 80.27; H 4.50; N 9.74. Calculated, %: $\text{C}_{20}\text{H}_{14}\text{N}_2\text{O}$, C 80.52; H 4.73; N 9.39.

General procedure for the synthesis of compounds 9-11

To a solution of 2-amino-3-carbonitrile 6-8 (1 mmol) in 1,4-dioxane (20 mL) was added triethylorthoformate (2 mmol) and acetic anhydride (2 mmol). The reaction mixture was heated under reflux condition for 2-4 h. After

the completion of the reaction, the solvent was removed and the precipitate was recrystallized from ethanol to afford the pure product 9-11.

Ethyl N-3-cyano-7,7-dimethyl-5-oxo-4-phenyl-5,6,7,8-tetrahydro-4H-chromen-2-ylformimidate (9): reaction time: 2h, Yield 85%, m.p 178-180 °C. IR (KBr): 2206, 1674 cm^{-1} . ^1H N.M.R. (CDCl_3) (ppm) (*J*, Hz): 1.11 (3H, s, CH_3), 1.19 (3H, s, CH_3), 1.28 (3H, t, CH_3 , *J*=6.6), 2.25 (2H, s, CH_2), 2.47 (2H, s, CH_2), 4.37 (2H, q, *J*=6.6, CH_2), 4.54 (1H, s, H_{pyran}), 7.21-7.35 (5H, m, $\text{H}_{\text{aromatic}}$), 8.25 (1H, s, H_{imine}). Elemental analysis. Found, %: C 72.27; H 6.14; N 7.89. $\text{C}_{21}\text{H}_{22}\text{N}_2\text{O}$, Calculated, %: C 71.98; H 6.33; N 7.99.

Ethyl N-3-cyano-7-hydroxy-4-phenyl-4H-chromen-2-ylformimidate (10): reaction time: 3h, Yield 80%, m.p 166-168 °C. IR (KBr): 3110, 2224 cm^{-1} . ^1H N.M.R. (CDCl_3) (ppm) (*J*, Hz): 1.34 (3H, t, *J*=5.6, CH_3), 4.39 (2H, q, *J*=5.6, CH_2), 4.83 (1H, s, H_{pyran}), 6.79-7.37 (8H, m, $\text{H}_{\text{aromatic}}$ and 1H, OH), 8.39 (1H, s, H_{imine}). ^{13}C N.M.R. (CDCl_3) (ppm): 21.1, 42.5, 64.2, 81.2, 110.2, 119.4, 119.7, 127.3, 127.7, 127.9, 128.1, 128.2, 129.0, 143.3, 157.0, 159.6, 169.1. Elemental analysis. Found, %: C 71.01; H 5.21; N 8.51. $\text{C}_{19}\text{H}_{16}\text{N}_2\text{O}_3$, Calculated, %: C 71.24; H 5.03; N 8.74.

Ethyl N-2-cyano-1-phenyl-1H-benzo[f]chromen-3-ylformimidate (11): reaction time: 4 h, Yield 75%, m.p 224-226 °C. IR (KBr): 2224 cm^{-1} . ^1H N.M.R. (CDCl_3) (ppm) (*J*, Hz): 1.35 (3H, t, *J*=7.1, CH_3), 4.40 (2H, q, *J*=7.1, CH_2), 5.30 (1H, s, H_{pyran}), 7.21-7.86 (11H, m, $\text{H}_{\text{aromatic}}$), 8.45 (1H, s, H_{imine}). ^{13}C N.M.R. (CDCl_3) (ppm): 13.9, 40.8, 64.2, 82.2, 113.9, 116.8, 118.3, 123.7, 125.2, 127.4, 127.7, 128.6, 129.1, 129.9, 130.7, 131.5, 143.1, 147.6, 156.6, 159.5, 168.2. Elemental analysis. Found, %: C 77.76; H 5.02; N 9.021. $\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}_2$, Calculated, %: C 77.95; H 5.12; N 9.03.

General procedure for the synthesis of compounds 12-20

A mixture of imine 9-11 (1 mmol) and ammonia or primary amine (1 mmol) in ethanol or 1,4-dioxane (15 mL) was refluxed for the indicated time (Table 1). After completion of the reaction, the solid material was separated and recrystallized from ethanol to give compounds 12-20.

4-Imino-8,8-dimethyl-5-phenyl-5,7,8,9-tetrahydro-3H-chromeno[2,3-*d*]pyrimidin-

6(4*H*)-one (12): IR (KBr): 3308-3464, 1691 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 0.90-1.10 (6H, s, 2CH₃), 2.14 (2H, s, CH₂), 2.35 (2H, s, CH₂), 5.58 (1H, s, H_{pyran}), 6.90-7.31 (6H, m, H_{aromatic}), 7.81 (1H, s, NH), 8.07 (1H, s, H_{imine}), 11.61 (1H, s, NH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 26.9, 28.5, 43.6, 50.5, 56.5, 127.0, 127.9, 128.5, 140.9, 143.5, 147.7, 156.7, 163.1, 164.4, 165.1, 193.8. Elemental analysis. Found, %: C 71.27; H 5.70; N 13.28. C₁₉H₁₉N₃O₂, Calculated, %: C 71.01; H 5.96; N 13.08.

4-Imino-8,8-dimethyl-3-(3-nitrophenyl)-5-phenyl-5,7,8,9-tetrahydro-3*H*-chromeno[2,3-*d*]pyrimidin-6(4*H*)-one (13): IR (KBr): 3421, 1684, 1645 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 1.00 (3H, s, CH₃), 1.08 (3H, s, CH₃), 2.13 (2H, s, CH₂), 2.64 (2H, s, CH₂), 4.40 (1H, s, H_{pyran}), 7.21-9.00 (11H, m, H_{aromatic}, H_{imine}, NH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 25.1, 25.4, 27.3, 29.8, 31.2, 52.6, 101.2, 123.7, 124.9, 126.1, 128.7, 129.9, 131.2, 133.7, 135.1, 139.5, 142.1, 143.9, 145.1, 148.7, 153.2, 159.9, 173.4. Elemental analysis. Found, %: C 67.62; H 4.91; N 12.50. C₂₅H₂₂N₄O₄, Calculated, %: C 67.86; H 5.01; N 12.66.

4-Imino-5-phenyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol (14): IR (KBr): 3462, 3310, 3094 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 5.15 (1H, s, H_{pyran}), 6.51-7.25 (10H, m, H_{aromatic}, NH), 8.09 (1H, s, H_{imine}), 9.65 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 37.9, 96.6, 103.5, 112.8, 115.9, 127.1, 127.7, 129.0, 130.2, 145.4, 150.5, 156.9, 157.7, 162.9, 163.1. Elemental analysis. Found, %: C 70.23; H 4.63; N 14.59. C₁₇H₁₃N₃O₂, Calculated, %: C 70.09; H 4.50; N 14.42.

3-(2-Chlorophenyl)-4-imino-5-phenyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol (15): IR (KBr): 3393, 3157, 1635 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 5.48 (1H, s, H_{pyran}), 6.56-8.26 (14H, m, H_{aromatic}, NH, H_{imine}), 9.79 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 37.7, 98.9, 103.5, 113.1, 115.3, 126.8, 127.5, 127.9, 128.6, 129.3, 129.9, 130.4, 136.2, 144.8, 150.1, 156.7, 157.8, 159.82, 159.83, 163.3. Elemental analysis. Found, %: C 68.89; H 4.21; Cl 8.61; N 10.29. C₂₃H₁₆ClN₃O₂, Calculated, %: C 68.74; H 4.01; Cl 8.82; N 10.46.

4-Imino-3-(3-nitrophenyl)-5-phenyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol

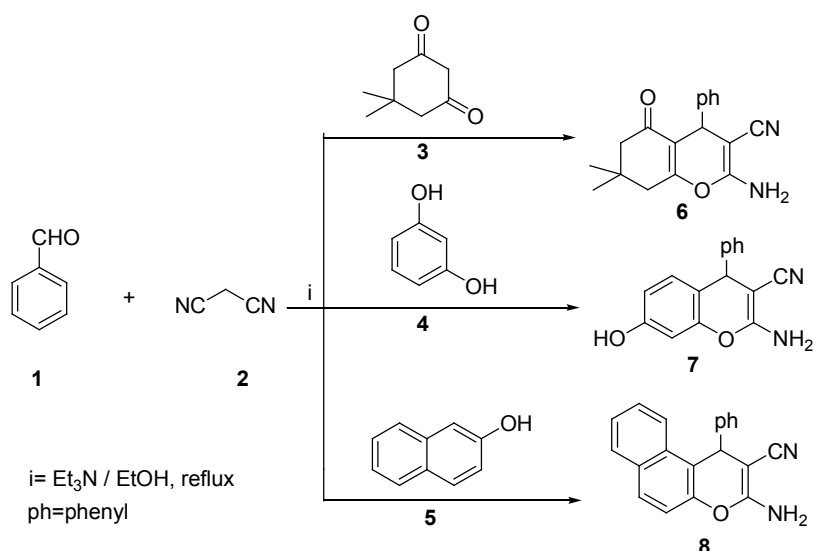
(16): IR (KBr): 3402, 3142, 1612 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 5.74 (1H, s, H_{pyran}), 6.60-8.05 (12H, m, H_{aromatic}), 8.60 (1H, s, H_{imine}), 8.95 (1H, s, NH), 9.76 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 37.2, 100.4, 103.7, 107.6, 113.3, 115.3, 117.6, 120.4, 127.3, 127.5, 129.3, 130.1, 130.2, 141.2, 145.3, 148.3, 150.3, 156.5, 158.0, 159.0, 163.6. Elemental analysis. Found, %: C 66.76; H 3.84; N 13.35. C₂₃H₁₆N₄O₄, Calculated, %: C 66.99; H 3.91; N 13.59.

4-Imino-5-phenyl-3-*p*-tolyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol (17): IR (KBr): 3427, 3092, 1635 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 2.24 (3H, s, CH₃), 5.63 (1H, s, H_{pyran}), 6.56-7.41 (12H, m, H_{aromatic}), 8.29 (1H, s, H_{imine}), 8.31 (1H, s, NH), 9.71 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 20.9, 37.2, 99.0, 103.5, 113.1, 115.5, 122.1, 127.3, 127.6, 129.2, 129.3, 130.2, 132.7, 137.2, 145.4, 150.3, 156.6, 157.8, 159.4, 163.1. Elemental analysis. Found, %: C 75.69; H 4.91; N 11.26. C₂₄H₁₉N₃O₂, Calculated, %: C 75.57; H 5.02; N 11.02.

3-(4-Ethylphenyl)-4-imino-5-phenyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol (18): IR (KBr): 3421, 3109, 1635 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 1.13 (3H, t, *J*=7.0, CH₃), 2.49 (2H, q, *J*=7.0, CH₂), 5.63 (1H, s, H_{pyran}), 6.03-7.42 (12H, m, H_{aromatic}), 8.29 (1H, s, H_{imine}), 8.35 (1H, s, NH), 9.76 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 16.2, 28.0, 37.1, 99.0, 103.5, 113.1, 115.5, 122.1, 127.3, 127.5, 128.1, 129.2, 130.3, 137.3, 139.3, 145.4, 150.3, 156.6, 157.8, 159.4, 163.1. Elemental analysis. Found, %: C 75.79; H 5.41; N 10.51. C₂₅H₂₁N₃O₂, Calculated, %: C 75.93; H 5.35; N 10.63.

3-Benzyl-4-imino-5-phenyl-4,5-dihydro-3*H*-chromeno[2,3-*d*]pyrimidin-8-ol (19): IR (KBr): 3340, 3115 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm) (*J*, Hz): 4.36 (1H, s, H_{benzyl}), 4.78 (1H, s, H_{pyran}), 6.54-7.47 (13H, m, H_{aromatic}), 7.79 (1H, s, NH), 8.51 (1H, s, H_{imine}), 9.32 (1H, s, OH). ^{13}C N.M.R. (DMSO- d_6) (ppm): 42.9, 64.4, 121.2, 122.8, 123.7, 134.1, 136.5, 138.2, 140.5, 142.1, 144.3, 145.7, 148.3, 150.6, 152.1, 154.7, 156.7, 158.2, 159.6, 160.4. Elemental analysis. Found, %: C 75.48; H 5.28; N 10.90. C₂₄H₁₉N₃O₂, Calculated, %: C 75.57; H 5.02; N 11.02.

12-Phenyl-10*H*-benzo[*f*]chromeno[2,3-*d*]pyrimidine-11(12*H*)-imine (20): IR (KBr): 3437, 3317, 1647 cm^{-1} . ^1H N.M.R. (DMSO- d_6) (ppm)



Scheme 1. Synthetic pathway of compounds 6-8.

(*J*, Hz): 6.06 (1H, s, H_{pyran}), 7.08-8.24 (14H, m, H_{aromatic}, H_{imine}, NH). ¹³C N.M.R. (DMSO-d₆) (ppm): 34.7, 97.6, 118.1, 123.6, 125.3, 127.1, 127.5, 128.1, 128.9, 129.1, 129.8, 131.3, 144.2, 148.3, 156.8, 162.9. Elemental analysis. Found, %: C 77.64; H 4.50; N 12.99. C₂₁H₁₅N₃O, Calculated, %: C 77.52; H 4.65; N 12.91.

Antibacterial activities

We used the agar disk diffusion method for this purpose. Each chemically synthesized materials (5 mg) was solved in DMSO as a solvent and 100 μL of known concentration of the test compounds was introduced onto the disks (7 mm) and then allowed to dry. Then the disk was introduced onto the upper layer of the medium with the bacteria. 100 μL of solvent (DMSO) was added to another disk and implanted as a negative control on each plate along with the standard drugs. The plates were incubated overnight at 37 °C. The inhibition zones were measured and compared with the standard drugs. The results are given in Table 2. The inhibition zone numbers are the average of three times of dependent experiments.

Results and Discussion

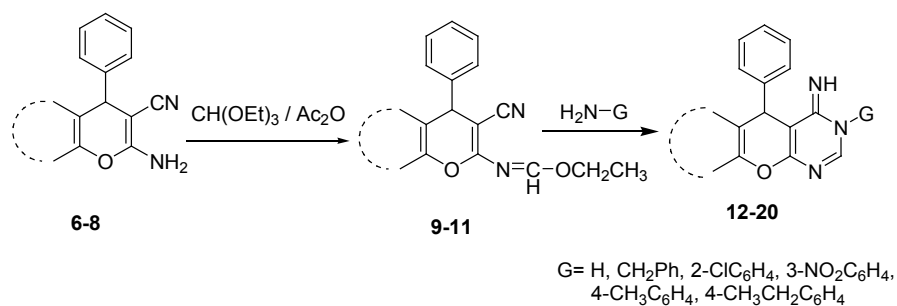
Compounds 6-8 were synthesized according to literature (33-35) (Scheme1). The appearance

of the IR absorptions bonds due to the NH₂ and CN groups of synthesized compounds 6-8 clearly confirmed the formation of these compounds.

The Imines 9-11 were prepared via reaction of 6-8 with triethylorthoformate in dioxane as a solvent. The IR spectra of 9-11 revealed the absence of the amino group, which is in support of imine formation. In the ¹H NMR spectra of these compounds, the appearance of triplet and quartet signals at high field is attributed to the resonance of ethoxy group protons. The resonance of imine proton at low field is also a good evidence for formation of imine structures.

Synthesis of compounds 12-20 was achieved through the reaction of 9-11 with ammonia or primary amine in ethanol or dioxane as a solvent (Scheme 2). In this reaction, when the G group is hydrogen, the reaction may produce two amino or imino tautomers. However, the NMR evidence is consistent with imino form (36), which shows two broad separate signals for two different NH groups. Yield of products after recrystallization from ethanol was in the range of 65%-93% (Table 1). The NMR spectra, as well as the elemental analysis data of these compounds are consistent with the expected structures.

The verification of antibacterial screening data revealed that seven out of nine tested materials showed antibacterial activities against *Staphylococcus aureus* and *Escherichia Coli*



Scheme 2. Synthetic pathway of compounds 9-20.

Table 1. Synthesis of chromenopyrimidines 12-20.

product	structure	Time (h)	Yield ^a %	M.P (°C)
12		3	75	275
13		2	69	255 dp
14		4	91	295
15		3	75	280 dp
16		6	65	270 dp
17		3	89	263
18		4	93	290 dp
19		5	71	250 dp
20		3	86	286 dp

a: isolated yields, dp: decomposed
 ph = phenyl

Table 2. Bacterial inhibition zone around disks containing samples.

Escherichia Coli (mm)	Staphylococcus aureus (mm)	Compound
15 ± 0.2 mm	23 ± 0.2 mm	12
10 ± 0.2 mm	20 ± 0.2 mm	13
13 ± 0.1 mm	22 ± 0.2 mm	14
18 ± 0.2 mm	25 ± 0.1 mm	15
9 ± 0.1 mm	21 ± 0.3 mm	16
10±0.1 mm	--	17
17 ± 0.1 mm	18 ± 0.3 mm	18
--	24 ± 0.2 mm	19
14 ± 0.2 mm	--	20
--	--	DMSO
Gentamicin 12 mm	Gentamicin 19 mm	Standard drugs

Indicates resistance of bacteria to compounds.

bacteria (Table 2). The maximum and minimum activities against *Staphylococcus aureus* were related to materials No. 15 and 18, respectively and maximum and minimum activities against *Escherichia Coli* were related to materials No. 15 and 16, respectively.

In conclusion, we have synthesized a series of novel chromenopyrimidine derivatives in suitable yields and the biological activity of these materials was also investigated.

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References

- (1) Lagoja I M. Pyrimidine as constituent of natural biologically active compounds. *Chem. Biochem.* (2005) 2: 1-50.
- (2) Hurst EW and Hull RTJ. Two new synthetic substances active against viruses of the psittacosis-lymphogranulomatrachoma group. *Med. Pharm. Chem.* (1961) 3: 215-229.
- (3) Machon Z and Cieplik J. Synthesis of furo[3,4-d]pyrimidine derivatives via reaction of 4-methylpyrimidine-5-carboxylic acids with thionyl chloride. *Synt.* (1986) 2: 142-144.
- (4) Machon Z and Cieplik J. Synthesis and antineoplastic effects of furo[3,4-d]pyrimidine derivatives. *J. Pharmacol. Pharm.* (1988) 40: 201-208.
- (5) Mohamed NR, El-Saidi MM, Ali YM and Elnagdi MH. Utility of 6-amino-2-thiouracil as a precursor for the synthesis of bioactive pyrimidine derivatives. *Bioorg. Med. Chem.* (2007) 15: 6227-6235.
- (6) Cieplik J, Pluta J and Meler G. Synthesis and biological investigations of pyrimidine derivatives. *Arch. Pharm.* (1997) 330: 237-241.
- (7) Habib N S, Soliman R, El-Tombary A A, El-Hawash S A and Shaaban O G. Synthesis of thiazolo[4,5-d]pyrimidine derivatives as potential antimicrobial agents. *Arch. Pharmacol. Res.* (2007) 30: 1511-1520.
- (8) Cieplik J, Pluta J and Gubrynowicz O. Synthesis and antibacterial activity of new sulphonamides of pyrimidine. *Sci. Pharm.* (2000) 68: 333-341.
- (9) Cieplik J, Raginia M, Pluta J, Gubrynowicz O, Bryndal I and Lis T. Synthesis and antibacterial properties of 1,2,3-aryl-1,2,3,4-tetrahydropyrimido[4,5-d]pyrimidine derivatives. *Acta Pol. Pharm. Drug Res.* (2008) 65: 427-434.
- (10) Pecorari P, Rinaldi M, Costantino L, Provvisionato A, Cermelli C and Portolani M. Synthesis and biological activity of pyrimido[2,1-b][1,3]thiazine, [1,3]thiazino[3,2-a]purine and [1,2,3]triazolo[4,5-d][1,3]thiazino[3,2-a]pyrimidine derivatives and thiazole analogues. *Farmaco.* (1991) 46: 899-911.
- (11) Al-Thebeiti MS. Synthesis of some new derivatives of thiazolo[3,2-a] pyrimidine-3,5,7(2H)-trione of potential biological activity. *Boll. Chim. Farm.* (2001) 140: 221-223.
- (12) Tozkoparan B, Ertan M, Kelicen P and Demirdamar R. Synthesis and anti-inflammatory activities of some thiazolo[3,2-a]pyrimidine derivatives. *Farmaco.* (1999) 54: 588-593.
- (13) El-Gaby MSA, Abdel-Hamide SG, Ghorab MM and El-Sayed SM. Synthesis and anticancer activity *invitro* of some new pyrimidines. *Acta Pharm.* (1999) 49: 149-158.
- (14) Dave CG, Shah DR, Shah GK, Pandya PS, Dave KC and Patel VJ. Pyridopyrimidines part III. Synthesis and analgesic activity of 4-aminopyrido[2,3-d]pyrimidines.

- Indian J. Pharm. Sci.* (1986) 48: 75-77.
- (15) Stulik K and Pacakova V. High performance liquid chromatography of biologically important pyrimidine derivatives with ultraviolet-voltametric-polarographic detection. *J. Chromatogr.* (1983) 1: 77-86.
- (16) Dave CG, Shah PR, Desai VB and Srinivasan S. Synthesis and biological activity of some pyridylthioureas and pyridopyrimidinethiones. *Indian J. Pharm. Sci.* (1982) 44: 83-85.
- (17) Shigeta S, Mori S, Watanabe F, Takahashi K, Nagata T, Koike N, Wakayama T and Saneyoshi M. Synthesis and antiherpes virus activities of 5-alkyl-2-thiopyrimidine nucleoside analogues. *Antivir. Chem. Chemother.* (2002) 13: 67-82.
- (18) Mobinikhaledi A and Kalhor M. Synthesis and biological activity of some oxo- and thioxopyrimidines. *Int. J. Drug Dev. Res.* (2010) 2: 268-272.
- (19) Taylor E C and Patel H H. Synthesis of pyrazolo [3,4-d]pyrimidine analogues of the potent agent N-{4-[2-(2-amino-4[3H]-oxo-7H-pyrrolo[2,3-d]pyrimidin-5-yl)ethyl]benzoyl}-L-glutamic acid (LY231514). *Tetrahedron* (1992) 48: 8089-8100.
- (20) Hynes JB, Greatz RF and Ashton WT. Potential Purine Antagonists. I. Synthesis of some bis(2,4-diaminopyrimidines) and bis(2,4-diaminoquinazolines) as potential antimalarial agents. *J. Med. Chem.* (1972) 15: 1332-1333.
- (21) Robins R K. Potential purine antagonists. I. Synthesis of some 4,6-substituted pyrazolo[3,4-d]pyrimidines. *J. Am. Chem. Soc.* (1956) 78: 784-790.
- (22) White FR. 4-Aminopyrazolo[3,4-d]pyrimidine and three derivatives. *Cancer Chemother. Rep.* (1959) 3: 26-36.
- (23) Nekoeian AA, Khalili A, Javidnia K, Mehdipour AR and Miri R. Antihypertensive effects of some new nitroxyalkyl 1,4-dihydropyridines derivatives in rat model of two-kidney, one-clip hypertension. *Iran. J. Pharm. Res.* (2009) 8: 193-199.
- (24) Jianga N, Denga X, Lib F and Quana Z. Synthesis of novel 7-substituted-5-phenyl-[1,2,4]triazolo[1,5-a]pyrimidines with anticonvulsant activity. *Iran. J. Pharm. Res.* (2012) 11: 799-806.
- (25) Sasada T, Kobayashi F, Sakai N and Konakahara T. An unprecedented approach to 4,5-disubstituted pyrimidine derivatives by a ZnCl₂-catalyzed three-component coupling reaction. *Org. Lett.* (2009) 11: 2161-2164.
- (26) Munawar MA, Azad M, Siddiqui HL and Nasim FH. Synthesis and antimicrobial studies of some quinolinylpyrimidine derivatives. *J. Chin. Chem. Soc.* (2008) 55: 394-400.
- (27) Abd El-Fattah O, Hassan Abbas EM, Ahmed Mohamed N and Abd-Elmoez SI. Synthesis and evaluation of some tetrahydropyrimidine derivatives as antimicrobial. *Aust. J. Basic Appl. Sci.* (2010) 4: 27-36.
- (28) Khalafi Nezhad A, Zare A, Parhami A and Soltani Rad MN. Practical synthesis of some novel unsymmetrical 1,3-dialkyl pyrimidine derivatives at room temperature. *Arkivoc.* (2006) 12: 161-172.
- (29) Foroughifar N and Mobinikhaledi A. Synthesis and antibacterial activities of novel pyrimidines derived from 2-oxo(or thioxo)-4-phenyl-5-carbethoxy-6-methyl-1,2,3,4-tetrahydropyrimidine. *Asian J. Chem.* (2002) 14: 614-618.
- (30) Foroughifar N, Mobinikhaledi A and Fathinejad H. Microwave-assisted synthesis of some pyrimidine derivatives using polyphosphate ester (PPE) in ceramic bath. *Phosphorus, Sulfur Silicon Relat. Elem.* (2003) 178: 1241-1246.
- (31) Foroughifar N, Mobinikhaledi A and Fathinejad H. Synthesis of some Biginelli compounds in solvent medium using a photochemistry method. *Phosphorus, Sulfur Silicon Relat. Elem.* (2003) 178: 495-500.
- (32) Foroughifar N, Mobinikhaledi A, Karimi G and Foroughifar N. One-pot synthesis of tetrahydropyrimidines catalyzed by zeolite. *Synth. React. Inorg. Met-Org. Chem.* (2007) 37: 279-282.
- (33) Tahmassebi D, Bryson JA and Binz SI. 1,4-Diazabicyclo[2.2.2]octane as an efficient catalyst for a clean, one-pot synthesis of Tetrahydrobenzo[b]pyran derivatives via multicomponent reaction in aqueous media. *Synth. Commun.* (2011) 41: 2701-2711.
- (34) Shestopalov AM, Emelianova YM and Nesterov VN. One-step synthesis of substituted 2-amino-4H-chromenes and 2-amino-4H-benzo[f]chromenes. Molecular and crystal structure of 2-amino-3-cyano-6-hydroxy-4-phenyl-4H-benzo[f]chromene. *Russ. Chem. Bull.* (2002) 51: 2238-2243.
- (35) Heravi MM, Baghernejad B and Oskooie HA. A novel and efficient catalyst to one-pot synthesis of 2-amino-4H-chromenes by methanesulfonic acid. *J. Chin. Chem. Soc.* (2008) 55: 659-662.
- (36) Kitamura T, Hikita A, Ishikawa H and Fujimoto A. *Spectrochimica Acta Part A* (2005) 62: 1157-1164.

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